

SHORT
COMMUNICATIONSSynthesis of New Pyrazole Derivatives
from Benzylidenemalononitrile

S. B. Nosachev, N. A. Shchurova, E. A. Tyrkova, and A. G. Tyrkov

Astrakhan State University, pl. Shaumyana 1, Astrakhan, 414000 Russia

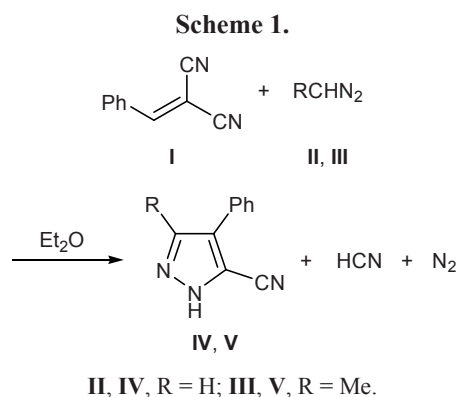
e-mail: tyrkov@rambler.ru

Received November 9, 2007

DOI: 10.1134/S107042800903021X

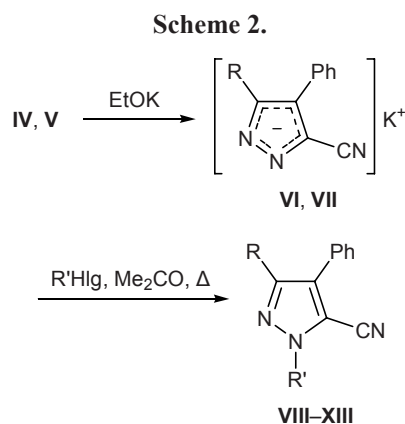
1,3-Dipolar cycloaddition of diazoalkanes to acetylenes [1] or styrenes [2] activated by electron-withdrawing substituents underlies a traditional procedure for the synthesis of pyrazoles. For example, 1-bromo-1-nitro-2-phenylethene reacts with diazomethane in diethyl ether to give 5-bromo-5-nitro-4,5-dihydro-1*H*-pyrazole, and treatment of the latter with hydrochloric acid or a solution of sodium hydrogen carbonate leads to the formation of 5-bromo- or 5-nitro-4-phenyl-1*H*-pyrazole, respectively [3].

With a view to study competing effects of functional groups in an analog of 1-nitro-2-phenylethene, benzylidenemalononitrile (**I**), we examined its reactions with diazomethane (**II**) and diazoethane (**III**). The cycloaddition of diazoalkanes **II** and **III** to dipolarophile **I** occurred under mild conditions (in diethyl ether at -5 to 5°C), and the products were the corresponding 4-phenyl-1*H*-pyrazole-5-carbonitriles **IV** and **V** (Scheme 1).



In the IR spectra of compounds **IV** and **V** we observed an absorption band at 3550 cm^{-1} due to stretching vibrations of the NH group, and the NH

signal appeared in their ^1H NMR spectra as a broadened singlet at δ 10.8–10.9 ppm. The presence of a labile hydrogen atom on the nitrogen in molecules **IV** and **V** opens prospects in further functionalization of these compounds, e.g., via alkylation of the corresponding potassium salts **VI** and **VII**. The latter were prepared *in situ* by treatment of 4-phenyl-1*H*-pyrazole-5-carbonitriles **IV** and **V** with potassium ethoxide. The alkylation of potassium salts **VI** and **VII** with chloromethyloxirane, phenacyl bromide, and 4-toluenesulfonyl chloride resulted in the formation of previously unknown alkylation products **VIII**–**XI** and N-sulfonyl derivatives **XII** and **XIII** (Scheme 2) whose structure was confirmed by spectral data and elemental analyses. The IR spectra of **X** and **XI** contained an absorption band at 1705 cm^{-1} due to stretching vibrations of the carbonyl group, while compounds **XII** and **XIII** displayed absorption bands at 1150 and 1300 cm^{-1} belonging, respectively, to antisymmetric and symmetric



VI, **VIII**, **X**, **XII**, **R** = H; **VII**, **IX**, **XI**, **XIII**, **R** = Me; **VIII**, **IX**, **R'** = oxiran-2-ylmethyl; **X**, **XI**, **R'** = PhCOCH_2 ; **XII**, **XIII**, **R'** = 4- $\text{MeC}_6\text{H}_4\text{SO}_2$; **Hlg** = Cl, Br.

stretching vibrations of the SO₂ group. The ¹H NMR spectra of **VIII–XIII** were consistent with the assumed structures, and they resembled those reported for structurally related compounds [4, 5]. Protons in the oxirane ring of compounds **VIII** and **IX** resonated in the ¹H NMR spectra at δ 2.65–3.07 ppm. The ¹H NMR spectra of phenacyl and sulfonyl derivatives **X–XIII** contained signals typical of methylene protons in the NCH₂CO fragment (δ 6.44–6.45 ppm; compounds **X**, **XI**) or protons in the *p*-tolyl substituent (**XII**, **XIII**).

Thus the described reactions open a synthetic route to functionally substituted pyrazoles having various pharmacophoric groups on the nitrogen atom, which are difficult to obtain by other methods.

Diazomethane [6], diazomethane [7], and phenacyl bromide [8] were prepared according to known procedures.

Reaction of benzylidenemalononitrile (I) with diazoalkanes II and III (general procedure). A solution of diazoalkane **II** or **III** was added under stirring to a solution of 8 mmol of benzylidenemalononitrile (**I**) in 30 ml of anhydrous diethyl ether at –5 to 5°C until nitrogen no longer evolved. The mixture was then kept for 24 h at 25°C and evaporated under reduced pressure, and the oily residue was subjected to chromatography in a 250×10-mm column charged with activated silica gel (Silicagel 100–400 μm) using carbon tetrachloride as eluent.

4-Phenyl-1H-pyrazole-5-carbonitrile (IV). Yield 44%, mp 154–156°C. IR spectrum, ν, cm⁻¹: 3550 (NH), 2230 (CN). ¹H NMR spectrum, δ, ppm: 7.35 m (H_{arom}), 7.58 s (CH), 10.9 br.s (NH). Found, %: C 70.74; H 4.03; N 24.58. C₁₀H₇N₃. Calculated, %: C 71.01; H 4.14; N 24.85.

3-Methyl-4-phenyl-1H-pyrazole-5-carbonitrile (V). Yield 46%, mp 162–164°C. IR spectrum, ν, cm⁻¹: 3550 (NH), 2230 (CN). ¹H NMR spectrum, δ, ppm: 2.35 s (CH₃), 7.34 m (H_{arom}), 10.8 br.s (NH). Found, %: C 7.95; H 4.78; N 22.74. C₁₁H₉N₃. Calculated, %: C 72.13; H 4.92; N 22.95.

1-Substituted 4-phenyl-1H-pyrazole-5-carbonitriles VIII–XIII (general procedure). A solution of 6 mmol of compound **IV** or **V** in 20 ml of ethanol was cooled to 0±5°C, 6 mmol of potassium ethoxide was added, the mixture was kept for 0.5 h at 5°C, and the precipitate was filtered off, washed with cold ethanol, and dried. Pyrazole potassium salt **VI** or **VII** thus obtained was dispersed in 100 ml of acetone, 6 mmol of chloromethyloxirane or phenacyl bromide or a solution

of 6 mmol of *p*-toluenesulfonyl chloride in ethanol was added, and the mixture was heated for 2 h under reflux. The mixture was then kept for 3 days at 25°C and evaporated under reduced pressure, the residue was treated with diethyl ether (3×10 ml), the extract was evaporated, and the residue was subjected to chromatography in a 500×10-mm column charged with activated silica gel (Silicagel, 100–400 μm) using benzene (compounds **VIII**, **IX**) or chloroform (**X–XIII**) as eluent.

1-(Oxiran-2-ylmethyl)-4-phenyl-1H-pyrazole-5-carbonitrile (VIII). Yield 62 %, mp 105–108°C. ¹H NMR spectrum, δ, ppm: 2.65 d (CH₂), 3.06 m (CH), 4.23 d (CH₂), 7.38 m (H_{arom}), 7.59 s (CH). Found, %: C 69.10; H 4.68; N 18.51. C₁₃H₁₁N₃O. Calculated, %: C 69.33; H 4.89; N 18.67.

3-Methyl-1-(oxiran-2-ylmethyl)-4-phenyl-1H-pyrazole-5-carbonitrile (IX). Yield 64%, mp 118–120°C. ¹H NMR spectrum, δ, ppm: 2.35 s (CH₃), 2.64 d (CH₂), 3.05 m (CH), 4.21 d (CH₂), 7.35 m (H_{arom}). Found, %: C 70.05; H 5.27; N 17.38. C₁₄H₁₃N₃O. Calculated, %: C 70.29; H 5.44; N 17.57.

1-(2-Oxo-2-phenylethyl)-4-phenyl-1H-pyrazole-5-carbonitrile (X). Yield 72%, mp 141–143°C. IR spectrum, ν, cm⁻¹: 2230 (CN). 1705 (C=O). ¹H NMR spectrum, δ, ppm: 6.44 s (CH₂), 7.33–7.45 m (H_{arom}), 7.58 s (CH). Found, %: C 75.05; H 4.34; N 14.42. C₁₈H₁₃N₃O. Calculated, %: C 75.26; H 4.53; N 14.63.

3-Methyl-1-(2-oxo-2-phenylethyl)-4-phenyl-1H-pyrazole-5-carbonitrile (XI). Yield 74%, mp 152–153°C. IR spectrum, ν, cm⁻¹: 2230 (CN), 1705 (C=O). ¹H NMR spectrum, δ, ppm: 2.34 s (CH₃), 6.45 s (CH₂), 7.36–7.44 m (H_{arom}). Found, %: C 75.56; H 4.77; N 13.74. C₁₉H₁₅N₃O. Calculated, %: C 75.75; H 4.98; N 13.95.

1-(4-Methylphenylsulfonyl)-4-phenyl-1H-pyrazole-5-carbonitrile (XII). Yield 71%, mp 172–174°C. IR spectrum, ν, cm⁻¹: 2230 (CN); 1300, 1150 (SO₂). ¹H NMR spectrum, δ, ppm: 2.32 s (CH₃), 7.11–7.52 m (H_{arom}), 7.57 s (CH). Found, %: C 62.94; H 3.86; N 12.83. C₁₇H₁₃N₃O₂S. Calculated, %: C 63.16; H 4.02; N 13.00.

3-Methyl-1-(4-methylphenylsulfonyl)-4-phenyl-1H-pyrazole-5-carbonitrile (XIII). Yield 75%, mp 180–183°C. IR spectrum, ν, cm⁻¹: 2230 (CN); 1300, 1150 (SO₂). ¹H NMR spectrum, δ, ppm: 2.32 s (CH₃), 2.35 s (CH₃), 7.12–7.54 m (H_{arom}). Found, %: C 63.86; H 4.27; N 12.31. C₁₈H₁₅N₃O₂S. Calculated, %: C 64.09; H 4.45; N 12.46.

The IR spectra were recorded on an IKS-29 spectrometer from solutions in chloroform with a concentration of 40 mg/ml (cell path length $l = 0.1$ mm). The ^1H NMR spectra were measured from solutions in acetone- d_6 on a Tesla BS-487C spectrometer (80 MHz) relative to hexamethyldisiloxane as internal reference. The progress of reactions and the purity of products were monitored by ascending TLC on Silufol UV-254 plates using acetone–hexane (2 : 3) as eluent; spots were visualized by treatment with iodine vapor.

REFERENCES

1. Ivanskii, V.I., *Khimiya geterotsiklicheskikh soedinenii* (Chemistry of Heterocyclic Compounds), Moscow: Vysshaya Shkola, 1978, p. 164.
2. Baran'ski, A. and Kelarev, V.I., *Khim. Geterotsikl. Soedin.*, 1990, p. 435.
3. Parham, W.E. and Bleasdale, J.L., *J. Am. Chem. Soc.*, 1951, vol. 73, p. 4664.
4. *Structure Determination of Organic Compounds: Tables of Spectral Data*, Pretsch, E., Bühlmann, P., and Afholter, C., Eds., Berlin: Springer, 2000, 3rd ed. Translated under the title *Opredelenie stroeniya organicheskikh soedinenii*, Moscow: Mir, 2006, p. 192.
5. Gordon, A.J. and Ford, R.A., *The Chemist's Companion*, New York: Wiley, 1972. Translated under the title *Sputnik khimika*, Moscow: Mir, 1976, p. 291.
6. *Organikum. Organisch-chemisches Grundpraktikum*, Berlin: Wissenschaften, 1976, 15th edn. Translated under the title *Organikum*, Moscow: Mir, 1979, vol. 2, p. 247.
7. James, A., Marshally, I., and Patridge, I., *J. Org. Chem.*, 1968, vol. 33, p. 4090.
8. Becker, H.G.O., *Organikum. Organisch-chemisches Grundpraktikum*, Berlin: Wissenschaften, 1964, 3rd ed. Translated under the title *Obshchii praktikum po organicheskoi khimii*, Moscow: Mir, 1965, p. 467.